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Chapter Review Chemical Bond Vocabulary

A hydrogen bond is a dipole - dipole attraction between a partially positive hydrogen atom and the unshared electron pair of a strongly electronegative atom such as O, N, or F. Unlike ionic or covalent bonds, in which electrons are given up or shared, the hydrogen bond is a weaker attraction.

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A bond that forms between atoms when they share electrons.
oxidation number. A positive or negative number that is assigned to an element to show its combining ability in a compound.
binary compound. A compound that is composed of only two elements.
polyatomic ion. A group of atoms with a positive or negative charge.

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Chemical bonding the joining of atoms to form new substances
Chemical bond a force of attraction that holds two atoms together
Valence electrons the electrons in the outermost energy level of an atom
Ionic bond the force of attraction between oppositely charged ions
Ions charged particles that form during chemical changes when one or more valence electrons

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VOCABULARY SHEET CH.#13 Chemical bonding

UNIT 4 - CHEMICAL BONDING VOCABULARY REVIEW. Unequal sharing of electrons due to electronegativity differences between bonding atoms. Equal sharing of electrons due to same elements bonding together (same electronegativities). Covalent bond resulting from sharing two pair of electrons. Covalent bond resulting from sharing three pair of electrons. With a few exceptions, all atoms bond so as to have eight valence electrons (to be like noble gases) allowing them to be more stable and have lower ...

UNIT 4 - CHEMICAL BONDING VOCABULARY REVIEW Flashcards ...

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Start studying Chapter 4: Atoms and Bonding Review. Learn vocabulary, terms, and more with flashcards, games, and other study tools.

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CHAPTER 6 REVIEW Chemical Bonding SECTION 1 SHORT ANSWER Answer the following questions in the space provided.

1. a A chemical bond between atoms results from the attraction between the valence electrons and of different atoms. (a) nuclei (c) isotopes (b) inner electrons (d) Lewis structures 2. b A covalent bond consists of (a) a shared electron.

6 Chemical Bonding

Start studying Chapter 16 Science "Atomic Structure and

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Chemical Bonds". Learn vocabulary, terms, and more with flashcards, games, and other study tools.

Chapter 16 Science "Atomic Structure and Chemical Bonds ...

an attraction between atoms that allows the formation of chemical substances that contain two or more atoms. The bond is caused by the electrostatic force of attraction between opposite charges, either between electrons and nuclei, or as the result of a dipole attraction.

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Chapter Review. Part A. Vocabulary Review (page 43) 1. chemical property (1/1) 2. chemical change (4/2) 3. conservation of mass (3/2) 4. physical change (4/2) 5. density (1/1) 6. physical property (1/1) 7. boiling point (1/1) 8. behavior (1/1) 9. volume (1/1) 10. melting point (1/1) 11. state (2/1) 12. appearance (1/1)

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Part B. Concept Review ...

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a chemical bond in which 2 atoms share a pair of valence electrons. molecule. a neutral group of atoms that are joined together by 1 or more covalent bonds. metallic bond. the attraction between a metal cation and the shared electrons that surround it. alloy.

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Quia - Chapter 6: Chemical Bonds Vocabulary Review

Chapter 6 - Chemical Bonding. Description. Vocabulary/Key Terms. Total Cards. 40. Subject. Chemistry. Level. Not Applicable. Created. 03/16/2004. Click here to study/print these flashcards. ... the chemical bond resulting from electrical attraction between large numbers of cations and anions: Term.

Chapter 6 - Chemical Bonding Flashcards

Chemical Bonds Class Rēwew Part AO Vocabulary Review
Directions: Complete the sentence by writing the correct terms in the blanks I. An atom that has gained or lost electrons is called a(n) 2. An atom is .30 A(n) atoms of those elements. when its outer energy level is filled with electrons. tells what elements make up a compound and the ratios ...

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Headlines ...

Chapter 5 REVIEW Word Bank acid base chemical bond chemical change chemical formula indicator ion pH physical change radicals subscript Vocabulary Review Choose the word or words from the Word Bank that best complete each sentence. Write the answers on a sheet of paper. 1. A(n) changes color in an acid or a base. 2.

Solved: Chapter 5 REVIEW Word Bank Acid Base Chemical Bond ...

Chemical Bonds Vocabulary Chemical Bond - The force that holds atoms or ions together as a unit. Ionic Bond - The force that holds cations and anions together. Ion - an atom or groups of atoms that has a positive or

Chemical Bond Vocabulary Review Answers

covalent bonding: sharing of electron pairs between two atoms:

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chemical bond: mutual attraction between the nuclei and valence electrons of different atoms that binds the atoms together: non-polar covalent bond: covalent bond where bonding electrons are shared equally by the bonded atoms, resulting in a balanced electrical charge: polar

Quia - Chemistry Chapter 6 Vocabulary Review

Best Chemistry: Chapter 8 Covalent Bonding Flashcards ...
Chapter 8 Review Covalent Bonding Vocabulary 1. Covalent bond - a bond formed by sharing of electrons between atoms 2. Molecule - a neutral group of atoms joined together by covalent bonds 3. Diatomic molecule - a molecule consisting of two atoms 4.

Chapter 8 Covalent Bonding Vocabulary Review

Chapter 10 "Chemical Quantities" Vocab. the SI unit representing 6.02×10^{23} representative particles of a substance. the

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temperature and pressure at which one mole of gas occupies a volume of 22.4 L. equal volumes of gases at the same temperature and pressure contain equal numbers of particles.
Chapter 10 Chemical Quantities Vocabulary Review ...

Chemical Vocabulary Review Answers

Vocabulary A substance formed from the ionic bond of a metal with a nonmetal. The chemical bond that binds metallic ions to nonmetallic ions by electrical attraction. A characteristic of ionic substances, such as salts, that readily shatter when struck a sharp blow.

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